

DESCRIPTION

PHOTOTHERMAL CONVERSION SPECTROSCOPIC ANALYSIS METHOD,
AND PHOTOTHERMAL CONVERSION SPECTROSCOPIC ANALYSIS
5 APPARATUS FOR CARRYING OUT THE METHOD

Technical Field

The present invention relates to a photothermal
conversion spectroscopic analysis method and apparatus
10 according to which exciting light is convergently
irradiated onto a sample to form a thermal lens, and
detecting light is also irradiated onto the sample, and
detecting light that has passed through the sample is
received, whereby a change in intensity of the detecting
15 light caused by refraction due to the thermal lens is
measured, and in particular relates to a photothermal
conversion spectroscopic analysis method and apparatus
that enable high-precision ultramicroanalysis to be
carried out through measurement in a very small space,
20 and moreover enable measurement to be carried out easily
in any chosen location.

Background Art

In recent years, spectroscopic analysis has come to
25 be widely used as a method for carrying out analysis or
measurement on semiconductors, biological samples,
various types of liquid sample, and so on. However, with
a conventional spectroscopic analysis method, in the case
of analyzing a very small amount of a substance or a very
30 small sample in a very small space, it has been necessary
to carry out the measurement in a vacuum environment.
Moreover, there has been a problem that the sample may be
damaged or destroyed upon using an electron beam or ion
beam.

35 Moreover, when handling an extremely small amount of

a sample in a solution, biological tissue, or the like, it is essential to use an optical microscope that enables analysis to be carried out with high precision and high spatial resolution. The only type of such microscope
5 actually used is a laser fluorescence microscope. The target of analysis is thus naturally limited to being a molecule that is fluorescent with a laser fluorescence microscope.

Due to this state of affairs, there have been
10 demands for an analysis method according to which a vacuum environment is not required, analysis can be carried out without contacting or risking damaging the sample, and moreover the target of analysis is not limited to being a fluorescent molecule, and analysis can
15 be carried out with high precision and high spatial resolution.

A photothermal conversion spectroscopic analysis method that uses a thermal lens effect brought about by photothermal conversion is attracting attention as an
20 analysis method that satisfies these demands.

This photothermal conversion spectroscopic analysis method uses a photothermal conversion effect in which light is convergently irradiated onto a sample, whereupon a solute in the sample absorbs the light, and hence the
25 temperature of the solvent is locally raised by thermal energy released due to the absorbed light, whereby the refractive index changes, and hence a thermal lens is formed.

FIG. 3 is a view useful in explaining the principle
30 of a thermal lens.

In FIG. 3, exciting light is convergently irradiated onto an extremely small sample via an objective lens of a microscope, whereby a photothermal conversion effect is brought about. For most substances, the refractive index
35 drops as the temperature rises. Consequently, in the

sample onto which the exciting light has been convergently irradiated, the refractive index drops, with the drop in the refractive index being larger closer to the center of the converged light, which is where the extent of the rise in temperature is largest; moving away from the center of the converged light toward the periphery, the extent of the rise in temperature becomes smaller due to thermal diffusion, and hence the drop in refractive index becomes smaller. Optically, the resulting refractive index distribution produces exactly the same effect as a concave lens, and hence the effect is referred to as the thermal lens effect. The size of the thermal lens effect, i.e. the power of the concave lens, is proportional to the optical absorbance of the sample. Moreover, in the case that the refractive index increases with temperature, a similar effect is produced, but conversely the thermal lens is convex.

In the photothermal conversion spectroscopic analysis method described above, thermal diffusion in the sample, i.e. change in the refractive index of the sample, is observed, and hence the method is suitable for detecting concentrations in extremely small samples.

A photothermal conversion spectroscopic analysis apparatus that carries out the photothermal conversion spectroscopic analysis method described above is described, for example, in Japanese Laid-open Patent Publication (Kokai) No. 10-232210.

In such a photothermal conversion spectroscopic analysis apparatus, the sample is disposed below the objective lens of a microscope, and exciting light of a predetermined wavelength outputted from an exciting light source is introduced into the microscope, and thus convergently irradiated via the objective lens onto an extremely small region in the sample. A thermal lens is thus formed centered on the focal position of the

convergently irradiated exciting light.

Moreover, detecting light having a wavelength different to that of the exciting light is emitted from a detecting light source, and is introduced into the microscope, before exiting from the microscope. The detecting light that has exited from the microscope is convergently irradiated onto the thermal lens that has been formed in the sample by the exciting light. Upon passing through the sample, the detecting light is diverged or converged by the effect of the thermal lens. The diverged or converged detecting light exiting from the sample is taken as signal light, and passes through a converging lens and a filter, or just a filter, before being received by a detector and thus detected. The intensity of the detected signal light depends on the refractive index of the thermal lens formed in the sample.

The frequency of the detecting light may be the same as that of the exciting light, or the exciting light may also be used as the detecting light. In general, good sensitivity is obtained in the case that the exciting light and the detecting light are made to have different frequencies to one another.

However, with a photothermal conversion spectroscopic analysis apparatus as described above, the structure of the optical system and so on for the light sources, the measurement section and the detection section (photoelectric conversion section) is complex, and hence such an apparatus has been large in size and has thus lacked portability. Consequently, there is a problem in that when carrying out analysis or a chemical reaction using such a photothermal conversion spectroscopic analysis apparatus, there are limitations with regard to the installation site of the apparatus and the operation of the apparatus, and hence there is a problem of the work efficiency for a user being poor.

In many cases of using a photothermal conversion spectroscopic analysis method that makes use of a thermal lens, it is necessary for the focal position of the exciting light and the focal position of the detecting light to be different to one another. FIGS. 4A and 4B are views useful in explaining the formation position of a thermal lens and the focal position of detecting light in the direction of travel of exciting light; FIG. 5A shows a case in which an objective lens has chromatic aberration, and FIG. 5B shows a case in which the objective lens does not have chromatic aberration.

In the case that the objective lens 130 has chromatic aberration, as shown in FIG. 4A, the thermal lens 131 is formed at the focal position 132 of the exciting light, and the focal position 133 of the detecting light is shifted by an amount ΔL from the focal position 132 of the exciting light; changes in the refractive index of the thermal lens 131 can thus be detected as changes in the focal distance of the detecting light. On the other hand, in the case that the objective lens 130 does not have chromatic aberration, as shown in FIG. 4B, the focal position 133 of the detecting light is almost exactly the same as the focal position 132 of the exciting light, i.e. the position of the thermal lens 131. As a result, the detecting light is not refracted by the thermal lens 131, and hence changes in the refractive index of the thermal lens 131 cannot be detected.

However, the objective lens of a microscope is generally manufactured so as not to have chromatic aberration, and hence for the reason described above, the focal position 133 of the detecting light is almost exactly the same as the position of the thermal lens 131 formed at the focal position 132 of the exciting light (FIG. 4B). Changes in the refractive index of the

thermal lens 131 thus cannot be detected. There is thus a problem in that the position of the sample where the thermal lens is formed must be shifted from the focal position 133 of the detecting light every time

5 measurement is carried out as shown in FIG. 5A or 5B, or else the detecting light must be diverged or converged slightly using a lens (not shown) before being introduced into the objective lens 130 so that the focal position 133 of the detecting light is shifted from the thermal
10 lens 131 as shown in FIG. 6; this requires time and effort, and hence the work efficiency for a user is poor.

It is an object of the present invention to provide a photothermal conversion spectroscopic analysis method that enables measurement to be carried out with high
15 sensitivity, and a small-sized photothermal conversion spectroscopic analysis apparatus that carries out the method.

Disclosure of the Invention

20 To attain the above object, in a first aspect of the present invention, there is provided a photothermal conversion spectroscopic analysis method having a convergent irradiation step of convergently irradiating exciting light and detecting light onto a sample using a
25 converging lens, and a measurement step of measuring a change in intensity accompanying deflection of the detecting light upon passing through a thermal lens produced through the convergent irradiation of the exciting light, characterized in that the exciting light
30 and the detecting light convergently irradiated in the convergent irradiation step have different frequencies to one another, and the converging lens satisfies a condition that a length of a shift in a focal position of the detecting light from a focal position of the exciting
35 light is in a range of 2 times to 30 times a confocal

length at the frequency of the exciting light.

To attain the above object, in a second aspect of the present invention, there is provided a photothermal conversion spectroscopic analysis method having a
5 convergent irradiation step of convergently irradiating exciting light and detecting light onto a sample using a converging lens, and a measurement step of measuring a change in intensity accompanying deflection of the detecting light upon passing through a thermal lens
10 produced through the convergent irradiation of the exciting light, characterized in that the exciting light and the detecting light have different frequencies to one another, and the converging lens satisfies a condition that a shift in a focal position of the detecting light
15 relative to a focal position of the exciting light is in a range of 2 times to 25 times a confocal length at the frequency of the exciting light.

In the first and second aspects of the present invention, the converging lens is preferably a rod lens.

20 To attain the above object, in a third aspect of the present invention, there is provided a photothermal conversion spectroscopic analysis apparatus comprising a converging lens for convergently irradiating exciting light and detecting light onto a sample, and measurement
25 means for measuring a change in intensity accompanying deflection of the detecting light upon passing through a thermal lens produced through the convergent irradiation of the exciting light, characterized in that the convergently irradiated exciting light and detecting
30 light have different frequencies to one another, and the converging lens satisfies a condition that a length of a shift in a focal position of the detecting light from a focal position of the exciting light is in a range of 2 times to 30 times a confocal length at the frequency of
35 the exciting light.

To attain the above object, in a fourth aspect of the present invention, there is provided a photothermal conversion spectroscopic analysis apparatus comprising a converging lens for convergently irradiating exciting
5 light and detecting light onto a sample, and measurement means for measuring a change in intensity accompanying deflection of the detecting light upon passing through a thermal lens produced through the convergent irradiation of the exciting light, characterized in that the
10 convergently irradiated exciting light and detecting light have different frequencies to one another, and the converging lens satisfies a condition that a length of a shift in a focal position of the detecting light from a focal position of the exciting light is in a range of 2
15 times to 25 times a confocal length at the frequency of the exciting light.

In the third and fourth aspects of the present invention, the converging lens is preferably a rod lens.

20 Brief Description of the Drawings

FIG. 1 is a view schematically showing the construction of a photothermal conversion spectroscopic analysis apparatus according to an embodiment of the present invention;

25 FIG. 2 is a graph showing an example of the relationship between ΔL and signal intensity in the photothermal conversion spectroscopic analysis method;

FIG. 3 is a view useful in explaining the principle of a thermal lens;

30 FIG. 4A is a view useful in explaining the formation position of a thermal lens and the focal position of detecting light in the direction of travel of exciting light, and shows a case in which an objective lens has chromatic aberration;

35 FIG. 4B is a view useful in explaining the formation

position of the thermal lens and the focal position of the detecting light in the direction of travel of the exciting light, and shows a case in which the objective lens does not have chromatic aberration;

5 FIG. 5A is a view useful in explaining the formation position of the thermal lens and the focal position of the detecting light in the direction of travel of the exciting light, and shows a case in which the thermal lens is formed on the objective lens side relative to the
10 focal position of the detecting light;

 FIG. 5B is a view useful in explaining the formation position of the thermal lens and the focal position of the detecting light in the direction of travel of the exciting light, and shows a case in which the thermal
15 lens is formed on the opposite side to the objective lens relative to the focal position of the detecting light; and

 FIG. 6 is a view useful in explaining a method of detecting changes in refractive index of a thermal lens in a conventional photothermal conversion spectroscopic
20 analysis apparatus.

Best Mode for Carrying Out the Invention

 A photothermal conversion spectroscopic analysis
25 method, and a photothermal conversion spectroscopic analysis apparatus that carries out the method, according to an embodiment of the present invention will now be described in detail with reference to the drawings.

 FIG. 1 is a view schematically showing the
30 construction of a photothermal conversion spectroscopic analysis apparatus according to an embodiment of the present invention.

 In FIG. 1, a chopper 112 for modulating exciting light to improve the S/N ratio of a thermal lens signal,
35 described later, is disposed close to an exciting light

source 111 in the optical path of the exciting light, which is emitted from the exciting light source 111. The modulated exciting light has its direction of travel changed by a mirror 114, and then passes through a
 5 dichroic mirror 113. Out of the two faces of the dichroic mirror 113, detecting light from a detecting light source 120 is made to be incident on the face on the opposite side to the face on which the exciting light is incident. The detecting light is reflected by the
 10 dichroic mirror 113, thus becoming coaxial with the exciting light, and then the exciting light and the detecting light are led to a lens 10 having a suitable amount of chromatic aberration.

The lens 10 is held by a movable holder 15. In the
 15 present embodiment, the lens 10 is a gradient index rod lens. Note, however, that so long as the lens 10 has a predetermined amount of chromatic aberration, the lens 10 is not limited to a gradient index rod lens.

The rod lens 10 is comprised of a transparent
 20 cylindrical body made, for example, of glass or plastic, wherein the refractive index changes continuously from the center toward the periphery (see, for example, Japanese Examined Patent Application Publication (Kokoku) No. 63-63502).

25 Such a rod lens 10 is known as a converging light-transmitting body for which the refractive index $n(r)$ at a position a distance r from the central axis in the radial direction is given approximately by the quadratic equation in r ,

$$30 \quad n(r) = n_0 \{1 - (g^2/2)r^2\},$$

wherein n_0 represents the refractive index at the central axis, and g represents a quadratic distribution constant.

If the length z_0 of the rod lens 10 is chosen to be
 35 in a range of $0 < z_0 < \pi/2g$, then even though the rod

lens 10 has flat end faces, the image formation characteristics of the rod lens 10 will be the same as those of an ordinary convex lens; when a parallel light beam is incident on the rod lens 10, a focal point will
 5 be formed at a position a distance s_0 from the end of the rod lens 10 from which the light beam exits, where

$$s_0 = \cot(gz_0)/n_0g.$$

The rod lens 10 may be manufactured, for example, using the following method.

10 A rod is first formed from a glass having as principal components thereof 57 to 63 mol% of SiO_2 , 17 to 23 mol% of B_2O_3 , 5 to 17 mol% of Na_2O , and 3 to 15 mol% of Tl_2O , and then the glass rod is treated in an ion exchange medium such as a potassium nitrate bath, thus
 15 carrying out ion exchange between thallium ions and sodium ions in the glass and potassium ions in the medium, to give a refractive index distribution in the glass rod in which the refractive index decreases continuously from the center of the glass rod toward the periphery of the
 20 glass rod.

Because the optical axis of the cylindrical rod lens 10 intersects the two end faces of the rod lens 10 at right angles, the optical axis of the exciting light and the detecting light can easily be aligned with the
 25 optical axis of the rod lens 10. Furthermore, the rod lens 10 is considerably smaller in size than a microscope objective lens, and hence the apparatus as a whole can be made smaller in size.

Below the rod lens 10, a channel-formed plate-shaped
 30 member 20 through which a sample to be subjected to detection is made to flow is provided on an X-Y sample stage 125. The X-Y sample stage 125 can be moved on a plane orthogonal to the plane of the paper in FIG. 1.

The channel-formed plate-shaped member 20 through
 35 which the sample to be subjected to detection is made to

flow is comprised of three glass substrates 201, 202 and 203 bonded on top of one another; a channel 204 for carrying out mixing, agitation, synthesis, separation, extraction, detection or the like on the sample is formed
5 in the middle glass substrate 202.

From the perspective of durability and chemical resistance, the channel-formed plate-shaped member 20 is preferably made of a glass. In particular, envisaging usage with biological samples such as cell samples, for
10 example for DNA analysis, it is preferable to use a glass having high acid resistance and alkali resistance, specifically a borosilicate glass, a soda lime glass, an aluminoborosilicate glass, a quartz glass or the like. However, if the usage is limited accordingly, then a
15 channel-formed plate-shaped member 20 made of an organic substance such as a plastic can be used instead.

The focal position of the exciting light focussed by the rod lens 10 must be within the channel 204 of the channel-formed plate-shaped member 20. However, the rod
20 lens 10 does not have to be in contact with the channel-formed plate-shaped member 20. In the case that the rod lens 10 is made to be in contact with the channel-formed plate-shaped member 20, the focal position of the rod lens 10 can be adjusted through the thickness of the
25 upper glass substrate 201 of the channel-formed plate-shaped member 20. In the case that the thickness of the upper glass substrate 201 is insufficient, a spacer for adjusting the focal position may be inserted between the rod lens 10 and the upper glass substrate 201. In these
30 cases, subsequent adjustment of the focal position becomes unnecessary, and hence the apparatus can be made yet smaller in size.

A description will be now given regarding what should be the size of the shift (ΔL) between the focal
35 position of the exciting light and the focal position of

the detecting light. In the case that the target of measurement is an extremely thin film, a result has been obtained that $\sqrt{3}$ times the confocal length of the objective lens used to convergently irradiate the exciting light is optimum (Analyst, August 1995, Vol. 120, 2053). The confocal length I_c (nm) is given by $I_c = \pi \times (d/2)^2 / \lambda_1$. Here, d represents the Airy disk and is given by $d = 1.22 \times \lambda_1 / NA$, λ_1 represents the wavelength (nm) of the exciting light, and NA represents the numerical aperture of the rod lens 10 used. The value of ΔL represents the difference between the focal position of the exciting light and the focal position of the detecting light; there is no distinction according to whether the focal distance of the detecting light is longer or shorter than the focal distance of the exciting light.

However, the above optimum value for ΔL is only for the case that the exciting light and the detecting light have the same frequency, and the thickness of the sample does not exceed the confocal length of the objective lens.

At present, integration technology for carrying out chemical reactions in very small spaces is attracting attention from the perspective of the rapidity of chemical reactions, and carrying out reactions using very small amounts, on-site analysis and so on, and research is being carried out with vigor throughout the world.

As one example of such integration technology, there are apparatuses that aim to carry out any of mixing, reaction, separation, extraction, and detection on a sample in a very fine channel formed in a small glass substrate or the like. Such an apparatus may be used with a single function, for example for only separation, or may be used with a combination of functions.

As an example of an apparatus for only separation, an electrophoresis apparatus for analyzing extremely

small amounts of proteins, nucleic acids or the like has been proposed. Such an apparatus has a channel-formed plate-shaped member comprised of two glass substrates joined together (see, for example, Japanese Laid-open
5 Patent Publication (Kokai) No. 8-178897).

Since it is necessary to make the sample solution flow while maintaining the liquid characteristics thereof, the channel formed in the plate-shaped member used in such an apparatus usually has a depth of approximately 50
10 to 100 μ m. If photothermal conversion spectroscopic analysis is carried out with the solution that is the target of measurement flowing through such a channel, then the thickness (depth) of the target of measurement will be much greater than the confocal length of the
15 exciting light. For example, the confocal length in the case of converging exciting light of wavelength 532nm using an objective lens of NA (numerical aperture) 0.4 is 3.9 μ m, but the thickness of the channel is more than 10 times this. Comparing such a case in which the thickness
20 of the target of measurement is greater than the confocal length with the case of a thin film described above, the state will be as if a large number of thin films each having a thermal lens formed therein are piled on top of one another in the thickness direction, and hence
25 ultimately the effect will be the integral thereof; it is thus anticipated that the optimum value of the shift in the focal position between the exciting light and the detecting light will be larger than in the case of a thin film. However, if the shift in the focal position
30 between the exciting light and the detecting light is too large, then the amount of the detecting light passing through the thermal lens produced by the exciting light will be too low, and hence the detection sensitivity will drop. Regarding the chromatic aberration possessed by
35 the objective lens used in the photothermal conversion

spectroscopic analysis method, the shift (ΔL) between the focal position of the exciting light and the focal position of the detecting light is thus preferably in a range of 2 to 30 times, more preferably 2 to 25 times, yet more preferably 3 to 25 times, the confocal length for the exciting light.

In the case, for example, that the intensity of the exciting light used in the photothermal conversion spectroscopic analysis method is low, or the concentration of the target of measurement is low, the power of the thermal lens at a place away from the focal position of the exciting light will be low, and hence it is anticipated that the thermal lens effect integrated over the whole thickness of the target of measurement will be low. In such a case, it is preferable to make ΔL lower than stated above. ΔL is thus preferably in a range of 2 to 25 times, more preferably 3 to 25 times, yet more preferably 3 to 20 times, the confocal length for the exciting light.

An example will now be given of how much chromatic aberration can be obtained using a gradient index rod lens. An SLH lens as listed in the SELFOC lens catalog of Nippon Sheet Glass Co., Ltd. will be used as the gradient index rod lens. The lens characteristics at a diameter of 1.8mm are listed in the catalog, and hence these will be used converted to characteristic values for a diameter of 1mm.

In the case that the channel-formed plate-shaped member is made of Pyrex (registered trademark) glass, the thickness of the portion above the channel (i.e. the thickness of the upper glass 201) is 0.18mm, the depth of the channel is 0.1mm, the diameter of the SLH gradient index rod lens is 1mm, the effective diameter for light actually passing through the lens is 0.7mm, the rod length is 1.7mm, the wavelength of the exciting light is

488nm, the wavelength of the detecting light is 633nm, and the focal position of the exciting light is made to be in the very middle of the channel, the shift (ΔL) in the focal position between the exciting light and the
5 detecting light is 45 μ m. The NA at the focal position in this case is 0.46, and hence the confocal length for the exciting light is 2.7 μ m. ΔL is thus approximately 17 times the confocal length.

FIG. 2 is a graph showing an example of the
10 relationship between ΔL and the signal intensity in the photothermal conversion spectroscopic analysis method. The signal intensity shown in FIG. 2 was measured under the following conditions.

A microscope objective lens having no chromatic
15 aberration and an NA of 0.4 was used as an objective lens. The exciting light was led to the objective lens as it is, and the detecting light was diverged or converged before being introduced into the objective lens, thus changing the focal position of the detecting light. The sample
20 was an aqueous solution of Sunset Yellow of concentration 10^{-4} M, and this was put into a channel of thickness 0.1mm. The wavelength of the exciting light was 532nm, and the wavelength of the detecting light was 633nm. Under these conditions, the focal position of the exciting light was
25 made to be in the center of the channel, and the signal intensity obtained using the photothermal conversion spectroscopic analysis method was measured while shifting the focal position of the detecting light from the center of the channel along the optical axis; FIG. 2 shows a
30 plot of the results obtained.

According to FIG. 2, the signal intensity is highest when ΔL is approximately 60 μ m, which is approximately 15 times the confocal length for the exciting light (the confocal length for the exciting light was 3.9 μ m). The
35 range of ΔL over which the signal intensity obtained was

at least half of the maximum intensity is from 4 times to 27 times the confocal length for the exciting light.

A wavelength filter 116 that separates the exciting light and the detecting light and selectively transmits only the detecting light, and a photoelectric converter 117 that detects the detecting light that has been transmitted by the wavelength filter 116, are disposed at a location facing the rod lens 10 with the channel-formed plate-shaped member 20 therebetween, this being a location facing onto the channel 204 of the channel-formed plate-shaped member 20. A pinhole may be inserted in front of the photoelectric converter 117 so that only part of the detecting light is transmitted. The signal obtained by the photoelectric converter 117 is amplified by a pre-amplifier 121, and is then sent to a lock-in amplifier 122 where the signal is synchronized with the chopper 112, before being analyzed by a computer 123.

According to the present embodiment, the rod lens 10 has an amount of chromatic aberration suited to the wavelengths of the exciting light and detecting light used, and the dimensions of the channel 204 of the channel-formed plate-shaped member 20 used in the measurement; measurement can thus be carried out with high sensitivity, and moreover it is not necessary to externally provide an optical system for adjusting the focal positions of the exciting light and the detecting light, and hence the apparatus can be made smaller in size.

Industrial Applicability

As described in detail above, according to the present invention, the exciting light and the detecting light have different frequencies to one another, and the converging lens satisfies a condition that the length of the shift in the focal position of the detecting light

from the focal position of the exciting light is in a range of 2 times to 30 times the confocal length at the frequency of the exciting light; as a result, sufficient signal intensity can be obtained, and hence measurement
5 can be carried out with high sensitivity.

In the present invention, the exciting light and the detecting light have different frequencies to one another, and the converging lens satisfies a condition that the length of the shift in the focal position of the
10 detecting light from the focal position of the exciting light is in a range of 2 times to 25 times the confocal length at the frequency of the exciting light; as a result, the signal intensity is yet higher, and hence measurement can be carried out with yet higher
15 sensitivity.

In the present invention, the converging lens is a rod lens; as a result, an optical system for adjusting the focal positions of the exciting light and the detecting light can be omitted, and hence the apparatus
20 can be made smaller in size.